

Determination of Alkalinity (Carbonate and Bicarbonate hardness)





Use

The alkalinity of water is a measure of its capacity to neutralise acids. The alkalinity of natural water is due to the salts of carbonate, bicarbonate, borates, silicates and phosphates along with the hydroxyl ions in free state. However the major portion of the alkalinity in natural waters is caused by hydroxide, carbonate and bicarbonates with may be ranked in order of their association with high pH values. Alkalinity values provide guidance in applying proper doses of chemicals in water and waste water treatment processes, particularly in coagulation and softening.

The alkalinity of water (formerly also referred to as "p value" and "m value" or acid consumption) is defined as the consumption of hydronium ions up to pH 8.3 and pH 4.5. The results can be expressed as mg/L CaCO3,mmol/l °dH (Grad deutsche Härte/German hardness degree) and other units.

The Alkalinity method is stored as default method inside the TitroLine[®] 6000/7000/7750 titrators

Appliances

Titrator: TL 6000-M2/20 or TL 7000-M2/20 consists of

- Basic device
- Magnetic stirrer TM 235
- 20 mL exchange unit WA 20, with brown glass bottle for titrant complete
- pH combination electrode with integrated temperature sensor (A 162 2M-DIN-ID, N 1052 A, BL 18...)

Electrodes

- see above
- Calibration: DIN buffer pH= 4.00 and pH= 7.00



Reagents

- Titrant: H₂SO₄ or HCI 0.01 0.1 mol/l
- Titer: Possible with TRIS (Tris (hydroxymethyl)-aminomethan)

Description

Calibration

The pH combination electrode is calibrated in technical buffer pH=4.00 and pH= 7.00 or in DIN buffer pH= 4.01 and pH= 6.87.

Example of the calibration documentation:

		Calibration
Buffers used		
pH buffer 1:	TEC_4.000	
pH buffer 2:	TEC_7.000	
Measured values		
pH buffer 1:	TEC_4.000	165.6 mV / 23.4 °C
pH buffer 2:	TEC_7.000	-11.2 mV / 23.0 °C
Calibration data		
Slope:	99.4 % / -58.	8 mV/pH
Zero point:	pH 6.81 / -11.2 mV	
Temperature:	23.4 °C (a)	
Date and time:	07.03.13 / 15:04	

Determination of the exact concentration of the standard solution (option)

The exact concentration of the titrant can be determined using a titrimetric standard Tris (hydroxymethyl)-aminomethan. TRIS is dried in a desiccator overnight at room temperature before the titer determination is carried out.

The standard method for HCI/H2SO4 titrant ("titer HCI) is stored as a default method inside the TitroLine® 6000/7000/7750 titrators. With EDIT/F3 - Default method you can load this method. It is only necessary to change the name of the method:

Factor 2 (F2):	1000.0000
Factor 1 (F1):	1.0000

The factor 1 (F1) should be changed to 2 if as titrant 0.05 mol/l H_2SO_4 is used In a 100 or 150 ml, 0.2 to 0.3g TRIS are weighed accurately and dissolved in 60/80 mL of dist. water with stirring.



Documentation example for standard titration of H2SO4/HCI with TRIS



Calculation formula

Titre:	
Mol (M):	

(W*F2)/((EQ1-B)*M*F1) -> M103 121.14000

Weight (W): Blank value (B): Statistics:

man 0.0000 ml Off Factor 2 (F2): Factor 1 (F1): 1000.0000 1.0000



Method for standard titration of H2SO4/HCI with TRIS

Method data overall view

Method name:	Titre HCl	Created at:	09/13/12 14:23:02	
Method type:	Automatic titration	Last modification:	09/13/12 14:27:56	
Measured value:	pH	Damping settings:	None	
Titration mode:	Dynamic	Documentation:	GLP	
Dynamic:	Steep			
Measuring speed / drift:	Normal:	minimum holding time:	02 s	
		maximum holding time:	15 s	
		Measuring time:	02 s	
		Drift:	20 mV/min	
Initial waiting time:	0 s			
Titration direction:	Decrease			
Pretitration:	Off			
End value:	2.500 pH			
EQ:	On (1)			
Slope value:	Steep	Value:	700	

Dosing parameter			
Dosing speed:	100 %	Filling speed:	30 s
Maximum dosing volume:	50.00 ml		
Unit values			
Unit size:	20ml		
Unit ID:	10039005		
Reagent:	HCl 0.1 mol/L		
Batch ID:	no Charge		
Concentration [mol/l]:	0.10070		
Determined at:	12/05/11 19:18:45		
Expire date:	08/18/12		
Opened/compounded:	09/10/11		
Test according ISO 8655:	05/10/11		
Last modification:	09/13/12 14:35:18		
Device information			
Device: TitroLine 7000			
Serial number: 00012 Software version: 1230	Titre_HCl_13_09_12-14_36_21	pdf	

2/2



Titration of the sample

100 ml of water sample are pipetted into a beaker and titrated. If the consumption using HCl mol/l 0.1 should be less than 2 ml, the sample should be repeated using HCl 0.02 mol/l. Considering that this titration is an end-point titration, the titrator should be calibrated on a weekly or daily basis.

Select the Alkalinity method and start the titration.

Result example:







Method

The method is available as default method in the TL 6000/7000/TL 7750 titrator and ready to use.

N	Method data overall view			
	Method name: Method type: Measured value: Titration mode: Linear steps:	Alkalinity (p+m) Automatic titration pH End pt. 0.020 ml	Created at: Last modification: Damping settings: Documentation:	11/22/12 15:51:09 11/22/12 15:53:15 None GLP
	Measuring speed / drift:	Normal:	minimum holding time: maximum holding time: Measuring time: Drift:	02 s 15 s 02 s 20 mV/min
	Initial waiting time:	0 s		
	Titration direction:	Decrease		
	Pretitration:	Off		
	Endpoint 1:	pH 8.200	delta endpoint 1:	pH 0.500
	Endpoint 2:	pH 4.300	delta endpoint 2:	pH 1.000
	Enderster Et	P11 1000	Endpoint delay 2:	10 s
			Entertaine and a st	Acta.

Dosing parameter		
Dosing speed:	15 %	Filling speed:
Maximum dosing volume:	50.00 ml	
Unit values		
Unit size:	20ml	
Unit ID:	10039005	
Reagent:	HCl 0.1 mol/L	
Batch ID:	no Charge	
Concentration [mol/l]:	0.10030	
Determined at:	11/22/12 18:14:29	
Expire date:	08/18/12	
Opened/compounded:	09/10/11	
Test according ISO 8655:	05/10/11	

Last modification: 11/22/12 10:14:33

30 s



Notes

If you have any questions on the application, you can feel free to contact us..

优莱博技术 (北京)有限公司

